METABOLIC REACTIONS IN VITRO OF PSORALENS WITH LIVER AND EPIDERMIS*

BARBARA B. MANDULA and MADHU A. PATHAK

Harvard Medical School, Massachusetts General Hospital, Boston, MA 02114, U.S.A.

(Received 25 April 1977; accepted 19 October 1977)

Abstract—In recent years, psoralens have been widely-used clinically, but until now their in vitro and in vivo metabolic reactions have not been studied. The photochemotherapeutic agent 4,5',8-trimethylpsoralen (TMeP) is readily metabolized in vitro by mouse liver under mixed-function oxidase conditions, with formation of three products. One of the products is 4,8-dimethyl, 5'-carboxypsoralen which has been identified previously as an in vivo product of TMeP metabolism in the urine of mice and human subjects receiving oral TMeP. A second product, with a molecular weight of 244, is apparently 4,8-dimethyl, 5'-hydroxymethylpsoralen. A metabolic pathway that integrates in vivo and in vitro TMeP reactions is proposed. By contrast, 8-methoxypsoralen (8-MOP) apparently is minimally metabolized in vitro by mouse liver. Guinea pig epidermis did not show any in vitro reactions with either 8-MOP or TMeP. The difference in liver biotransformations of 8-MOP and TMeP may help explain why orally administered TMeP is less photosensitizing and less effective than 8-MOP in phototherapy of skin diseases such as psoriasis.

Certain photochemotherapeutic agents, commonly referred to as furocoumarins or psoralens, are used clinically in treating various skin diseases such as psoriasis [1–3], mycosis fungoides [4], vitiligo [5] and eczema [6]. Furocoumarins like 8-methoxypsoralen (8-MOP) and 4,5',8-trimethylpsoralen (TMeP) undergo photoconjugation reactions with pyrimidines [7,8]. The psoralens intercalate with double-stranded nucleic acids, and subsequent exposure to

8-Methoxypsoralen (8-MOP) 9-Methoxy-7H-furo [3,2-g] [1] benzopyran-7-one

4,5',8-Trimethylpsoralen (TMeP) 2,5,9-Trimethyl-7H-furo [3,2-g] [1] benzopyran-7-one

long wavelength ultraviolet radiation results in the formation of cyclobutane adducts and cross-links between two pyrimidines belonging to opposite strands of DNA [9,10]. Photoconjugation of psoralens to DNA has been established *in vivo* in several systems, including mammalian skin [11], human fibroblasts [12,13], *Escherichia coli* [9], and other biological systems [14,15]. The light-dependent conjugation of psoralens with epidermal DNA is

believed to inhibit DNA synthesis and cell division [1,16], thereby leading to clinical improvement of diseases such as psoriasis.

Although 8-MOP and TMeP have been used clinically, both orally and topically for the past 25 years, little is known of the metabolism of these photoactive compounds either in man or in laboratory animals [17]. Recently, we reported the isolation and identification of a major TMeP metabolite referred to as 4, 8-dimethyl, 5'-carboxypsoralen (DMe-CP) from the urine of mice and human volunteers receiving oral TMeP [18]. In the present study, we report on the transformation *in vitro* of TMeP and 8-MOP in mammalian liver and skin. TMeP is readily metabolized by the liver, and a metabolic pathway for its biotransformation is proposed.

MATERIALS AND METHODS

Materials

TMeP was tritiated by the New England Nuclear Corp. (Boston, MA). It was purified as described previously [11] and had a specific activity of $40 \,\mu\text{Ci}/\mu$ mole. [³H]TMeP was used routinely at about 0.3 $\mu\text{Ci}/\mu$ mole. Tritiated DMeCP was obtained from the urine of mice receiving [³H]TMeP as described recently [18]. Nonradioactive TMeP and 8-MOP were gifts from the Paul B. Elder Co. (Bryan, OH) and were used after repeated crystallization from ethanol. Ether extractions were performed with anhydrous ether from the Mallinckrodt Chemical Co. (St. Louis, MO). All other chemicals were reagent grade.

Tissue preparations

Male mice (strain CD-1 from the Charles River Breeding Laboratories) weighing 25-30 g were sacrificed by cervical dislocation. The liver or kidney was removed, weighed, and homogenized in a glass-Teflon Potter-Elvehjem homogenizer with 3 vol. of

^{*} This work was supported by NCI Grant 2-R01-CA 05003.

either Tris-sucrose buffer (0.25 M sucrose and 0.05 M Tris-chloride, pH 7.4) or phosphate sucrose buffer (0.25 M sucrose, 0.01 M sodium phosphate, pH 7.4, 0.001 M EDTA and 0.05 M nicotinamide).

Guinea pig epidermal homogenates were prepared by the procedure previously described [11]. Briefly, backs of female guinea pigs weighing 400–600 g were epilated 24-48 hr before the animals were sacrificed with ether anaesthesia. The epidermis was scraped off from the stretched skin and homogenized using 10 vol. of sucrose–phosphate buffer. To prepare microsomes, homogenates were centrifuged at $10,000 \, g$ for 20 min and the resulting supernatant fraction at $105,000 \, g$ for 60 min. Microsomes were resuspended in homogenizing buffer. Protein was determined by the procedure of Lowry ct al. [19] after precipitation by 5°_{\circ} trichloroacetic acid.

Reaction of tissue preparations with psoralens

The reaction mixture in 2.5 ml contained 50 mM sodium phosphate, pH 7.4; 3 mM glucose 6phosphate; 4 mM magnesium chloride; 0.16 mM NADP; 0.06 mM NAD; 0.4 mM EDTA; 12 mM nicotinamide; 0.3 units glucose 6-phosphate dehydrogenase; 350-450 nmoles TMeP or 8-MOP; and tissue preparation as indicated in table and figure legends. The reaction was standardly carried out at 37 for 30 min with vigorous shaking in 20 ml glass scintillation counting vials after a 5-min preincubation in the absence of the tissue preparation. The reaction was terminated by adding 10 ml anhydrous diethyl ether. In control reactions, the psoralen or tissue was added after ether. After vigorous shaking by hand, the other layer was removed and saved. Then 0.3 ml of 1 N HCl was added to the aqueous layer to give a pH of about 2, and the extraction was repeated with 10 ml of additional ether. The ether layers were combined, evaporated to dryness, and the residues dissolved in a small volume (0.2 ml) of ethanol for thin-layer chromatography (t.l.c.) and absorption spectroscopy.

The transformation *in vitro* of [⁸H]TMeP was quantitated by counting chromatograms. The transformation of nonradioactive 8-MOP or TMeP was measured by cluting the chromatograms with ethanol and obtaining absorption spectra. At 250 nm and a 1 cm light path, $10 \, \mu g$ TMeP/ml has an optical density of 1.4 or $E_{con}^1 = 3.20 \times 10^3$.

Thin-layer chromatography

Four solvent systems with the following composition by volume were used for t.l.c.: solvent A: $CHCl_3$ /ethyl acetate/acetic acid, 6/3/1; solvent B: $CHCl_3$ /ethyl acetate/acetic acid, 5/4/1; solvent C: toluene/acetic acid, 8/2; solvent D: H_2O only. Solvents A, B and C were used with Bakerflex Silica gel 1B or IB_2 plates with or without fluorescent indicator; solvent D was used on Bakerflex cellulose plates. Solvent fronts usually ran 12-15 cm. Compounds were detected by their characteristic fluorescence, color and R_1 values when viewed in the dark with long-wave (320-400 nm) or short-wave (254 nm) ultraviolet-emitting lamps.

Preparation of TMeP metabolites

Products M₁, M₂ and M₃ were prepared from standard TMeP incubations, using mouse liver

homogenate and t.l.c. $1B_2$ plates for chromatography in solvent A. Fluorescent substances M_1 and M_2 with R_1 values of 0.12 and 0.35, respectively, were cluted with ethanol along with corresponding bands from control t.l.c. plates. After ethanol evaporation, M_1 and M_2 were sometimes further purified by ether extraction from acidified aqueous solution.

Radioactivity determination

Tritium was counted in a Packard model 3300 Tri-Carb liquid scintillation counter using 10 ml Aquasol (New England Nuclear Corp.), Chromatograms were cut into thin strips which were placed in counting vials with 1 ml ethanol. Aquasol was added after shaking for 15 min. Counting efficiency was about 35 per cent. All data are corrected for background.

Spectroscopy

The mass spectra were taken on an AE1 MS-9 double focusing mass spectrometer with an electron impact source at 70 eV. The source temperature was the minimum required to obtain adequate ion intensity. Ultraviolet absorption of ethanolic solutions of the metabolites were obtained between 220 and 400 nm on a Bausch and Lomb Spectronic 505 recording spectrophotometer. The activation and emission spectra of ethanolic solutions of the metabolites were obtained on an Aminco-Keirs spectrophotophosphorimeter, model J-48201.

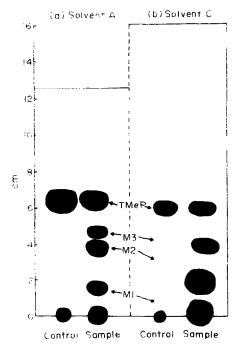


Fig. 1. Fluorescent spots on t.l.e. plates after reacting mouse liver homogenate with TMeP. Reaction samples were incubated for 30 min with homogenate equivalent to 75 mg fresh wt liver. For controls, liver was added after ether, i ther-extractable material was dissolved in 200 μ l of ethanol, and 20 μ l of ethanol solution was run on

the, plates in solvent A (a) and solvent C (b).

Table 1. Demonstration that mouse liver reaction with TMeP proceeds via mixed-function oxidase*

| Reaction mixture | Relative reaction (%) | | |
|-----------------------|-----------------------|------------|--|
| | Microsomes | Homogenate | |
| Complete | 100 | 100 | |
| Omit NAD | 82 | 82 | |
| Omit NADPH | 6 | 21 | |
| Omit NAD and NADPH | 3 | 3 | |
| Omit O ₂ † | 11 | 3 | |
| Omit liver | 0 | 0 | |

^{*} Incubations were conducted for 10 min with microsomes equivalent to 100 mg fresh wt liver, or for 30 min with homogenate equivalent to 50 mg liver.

RESULTS

Reaction in vitro of TMeP with mouse liver

TMeP metabolism in vitro was detected after incubating the drug with mouse liver homogenate, and separating ether-extractable substances by t.l.c. Three new fluorescent products clearly appeared in sample incubations, whereas control incubations showed unchanged substrate and a spot at the origin attributable to substances present in the liver extract (Fig. 1). With [³H]TMeP as substrate, new products were also confirmed as radioactive spots.

The transformation in vitro of TMeP by liver homogenate and microsomes appears to involve a mixed-function oxidase system (Table 1) [20]. Kinetics of the homogenate reaction with TMeP showed that the reaction rate was roughly linear with time and protein concentration until about 10 per cent of the initial TMeP had been metabolized (Fig. 2). Mouse liver homogenate reacted with TMeP at a rate of about 2.0 nmoles/mg of protein/min, or 320 nmoles/g fresh weight/min. Since TMeP is virtually insoluble in aqueous solution, we did not study the kinetic effects of changing its concentration.

Table 2. $R_{\rm f}$ values of metabolites of TMeP*

| Compound | $R_{\rm f}$ values | | |
|----------------|--------------------|-------|------|
| | A | С | D |
| M ₁ | 0.12 | 0.026 | 0.16 |
| M_2 | 0.35 | 0.12 | 0.13 |
| M_3 | 0.43 | 0.23 | 0.53 |
| DMeCP | 0.43 | 0.24 | 0.54 |
| TMeP | 0.57 | 0.37 | 0.0 |

^{*} Solvent systems: (A) CHCl₃/ethyl acetate/acetic acid, 6/3/1 by volume; (C) toluene/acetic acid, 8/2 by volume; and (D) water. Solvents A and C are run on Silica gel plates; D is run on cellulose plates.

Characterization of TMeP metabolites

The three fluorescent products, designated M_1 , M_2 and M_3 , formed when liver homogenate reacts with TMeP, were investigated further. Experiments were carried out to determine whether M_3 was DMeCP, since both substances had the same R_1 in solvent A (Table 2), and both DMeCP and M_3 exhibited blue fluorescence under short-wave ultraviolet light. By contrast, M_1 and M_2 exhibited white fluorescence.

 M_3 was isolated from an incubation mixture in vitro and rechromatographed in solvents A, C and D along with authentic DMeCP. Both M_3 and DMeCP have the same R_t values in the three solvents tested (Table 2). Furthermore, DMeCP and M_3 had the same ultraviolet absorption spectra with a characteristic double peak at 260 and 267 nm [18]. These results show that DMeCP is formed from TMeP after incubation in vitro with liver homogenate.

To characterize products M₁ and M₂, the compounds were prepared from reaction mixtures in vitro and chromatographed in solvents C and D. Ultraviolet absorption spectra of M₁ and M₂ were very similar to that of TMeP, with peaks at 250, 295 and 337 nm. The fluorescence spectrum of M₂ showed maximum excitation and emission wavelengths of 350 and 450 nm, respectively. Corresponding values for TMeP were 350 and 430 nm.

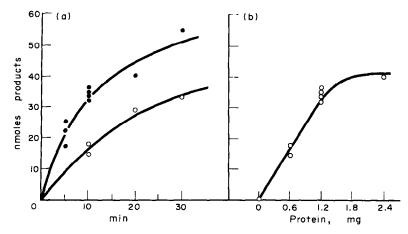
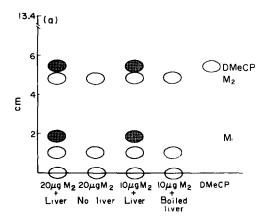


Fig. 2. Kinetics of TMeP reaction with mouse liver homogenate. Panel a: Reactions were carried out using 0.6 mg (⊙) or 1.2 mg protein (●). b: Reactions were incubated for 10 min with the indicated amount of protein.

[†] Nitrogen was bubbled through these samples, and activities were compared to samples through which air was bubbled.



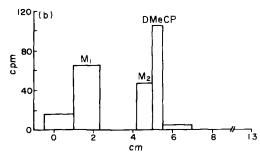


Fig. 3. Reaction of radioactive M_2 with mouse liver homogenate. Panel a: Indicated amounts of $[^3H]M_2$ (120 cpm/ μ g) were incubated for 30 min. with liver homogenate equivalent to 75 mg fresh wt. Aliquots of ether-extractable material were subjected to t.l.c. in solvent A. The observed fluorescent materials are schematically shown. The hatched areas show products formed. Panel b: Radioactivity distribution on the t.l.c. plate of liver reaction with 10 μ g $[^3H]M_2$.

Mass spectroscopy of the isolated M_1 and M_2 revealed that M_1 has a molecular weight of 260, which is 32 higher than TMeP. Besides the M_1 parent ion at 260, other major ions were detected with mass numbers 243, 232, 231, 215 and 187. M_2 showed a molecular weight of 244, and additional major ions appeared with mass numbers 227, 215 and 199. Mass spectra of the t.l.c. eluates obtained from corresponding control bands lacked the ions listed above.

Interrelationship of TMeP metabolites

The metabolites M₁, M₂ and DMeCP were each tested as substrates in a standard reaction mixture in vitro containing mouse liver homogenate and about 50 nmoles substrate. M2 was transformed both to DMeCP and to M₁, based on t.l.c. results in solvent A (Fig. 3) and solvent C (not shown). The experiments in Fig. 3 were repeated at least twice with cold M2 as well as [3H]M2 as substrate, and the data shown are reproducible and significant. The DMeCP formed from M₂ also was eluted with ethanol and showed a characteristic DMeCP ultraviolet absorption spectrum. DMeCP was not detectably metabolized further. With [3H]DMeCP as substrate, control and sample t.l.c. plates showed identical radioactive profiles in solvents A and C, with 95 per cent recovery of starting radioactivity. M₁ also did not show any in vitro conversion.

It was of interest to examine whether M_1 or M_2 , as

Table 3. Recovery of 8-MOP and TMeP after incubation with mouse liver microsomes*

| | TMeP (nmoles) | 8-MOP (nmoles) |
|--|------------------|-------------------|
| Initial substrate | 700 | 920 |
| Ether-extracted material | 420 | 700 |
| Amount run on t.l.c. | 84 | 140 |
| Amount recovered from t.l.c. Amount recovered from t.l.c. as | 52 | 105 |
| products Amount recovered from t.l.c. as | 32 | <4 |
| substrate | 20 | 105 |
| Per cent of recovered materials as products | 60% | < 5° 0 |

* Microsomes from 1 g weight of liver were incubated for 2 hr in 5 ml of reaction mix. Ether-extractable material was dissolved in ethanol. A portion was run on t.l.c. in solvent A, and fluorescent material was cluted with 1.5 ml ethanol and roughly quantitated by ultraviolet absorption spectroscopy: 0.012 O.D.₃₀₀ = 1 nmole 8-MOP/ml; 0.011 O.D.₂₉₅ = 1 nmole TMeP/ml.

well as DMeCP, was present in the urine of mice and human volunteers receiving oral TMeP (2 mg/mouse; 40 mg/70 kg for humans). Urine was extracted with ether, DMeCP was precipitated, and the remaining ether solutions were subjected to t.l.c. in solvents A, C and D. We could not detect any fluorescent substances resembling M₁ or M₂ that were present in sample urines but absent from control urines.

Reaction of mouse liver fractions with 8-MOP

Incubation experiments were undertaken to see whether 8-MOP, like TMeP, was metabolized by mouse liver *in vitro*. Homogenate equal to 75 mg fresh weight was incubated in a standard reaction with or without 8-MOP. Thin-layer chromatography in solvent B showed that the amount of 8-MOP in the incubated sample appeared undiminished when compared visually with the unincubated control. When 40 nmoles 8-MOP instead of the usual 400 nmoles was incubated, again no products appeared.

We also investigated the ability of liver microsomes to react with 8-MOP (see Table 3 for experimental details). After t.l.c. on IB₂ F plates, sample incubations showed small amounts of two new fluorescent spots with R_t values of 0.20 and 0.32 in solvent A, and R_t values of less than 0.1 in solvent C. Non-fluorescent products were not detected.

The extent of liver microsomal reaction with 8-MOP and TMeP was compared. Data in Table 3 show that 8-MOP is apparently minimally transformed to ether-extractable metabolites, whereas TMeP is readily transformed. Essentially all material recovered from the 8-MOP incubation has the same R_t and ultraviolet absorption spectrum as 8-MOP, and is presumed to be 8-MOP. Fluorescent 8-MOP metabolites were not formed in quantities sufficient to obtain absorption spectra.

Reaction of psoralens with guinea pig epidermis and mouse kidney

Since psoralens are used topically in treating certain human skin diseases, we investigated whether

Fig. 4. Proposed scheme for *in vitro* and *in vivo* metabolism of TMeP. Reactions 1, 2 and 3 were detected *in vitro*; reactions 4 and 5 are speculative. Products DMeCP and proposed hydroxy-DMeCP were isolated from mouse urine [18]. Abbreviations: TMeP = 4,5',8-trimethylpsoralen; DMeHyMeP = 4,8-dimethyl, 5'-hydroxymethylpsoralen; DMeCP = 4,8-dimethyl,5'-carboxypsoralen.

epidermis could readily metabolize 8-MOP or TMeP. No reaction products were detected by t.l.c. when guinea pig epidermal tissue fractions (whole homogenate, 10,000 g supernatant fraction or microsomes) were incubated with 8-MOP or TMeP under mixed-function oxidase conditions. The assay with [³H]TMeP would have detected a reaction rate greater than 0.2 nmole/g fresh wt/min; for 8-MOP, the estimated sensitivity is about 2 nmoles/g fresh wt/min. Recovery studies indicated that non-ether-extractable metabolites were apparently not being formed by epidermal homogenate, since recovery of starting material was greater than 95 per cent.

The guinea pig liver 10,000 g supernatant fraction appeared similar to mouse liver fractions both in forming M_1 and M_2 from TMeP and in not reacting readily with 8-MOP, based on fluorescent spots seen on t.l.c. plates.

Mouse kidney homogenate and the 10,000 g supernatant fraction failed to form detectable products from [³H]TMeP or [³H]DMeCP in vitro.

DISCUSSION

Although 8-MOP and TMeP have been widely used clinically since 1952, very little is known of their in vivo metabolic reactions. We have studied the metabolism of these two important photochemotherapeutic agents, and our observations enable us to propose a possible pathway for TMeP biotransformation. The fate of 8-MOP, however, remains to be established. It seems likely that furocoumarins such as TMeP are metabolized primarily in the liver and that liver mixed-function oxidases that require oxygen and NADPH are involved in this metabolism.

Three fluorescent products (M₁, M₂ and M₃) have been isolated after *in vitro* reaction of liver fractions with TMeP. Based on recovery experiments with [³H]TMeP as a substrate, these three products appear to account for all the ether-extractable compounds that are formed from TMeP. Figure 4 shows a possible interrelationship of the various products obtained from *in vivo* and *in vitro* metabolism of TMeP. Product M₃ is DMeCP, an end product of TMeP metabolism which has been identified in urine of mice and humans receiving TMeP orally [18].

Based on the following facts, we propose that M₂ is 4,8-dimethyl, 5'-hydroxymethylpsoralen (DMeHy-

MeP): (a) M_2 is enzymatically transformed to DMeCP by liver homogenate; (b) the molecular weight of M_2 is 244, which is 16 more than the molecular weight of TMeP; and (c) the tricyclic psoralen ring structure in M_2 is apparently intact, since the absorption spectra of M_2 and TMeP are essentially the same. M_2 could be metabolized to DMeCP either by a mixed-function oxidase reaction or by a dehydrogenase using NAD or NADP as a co-factor. Additional experiments are necessary to distinguish these possibilities.

We propose that M_1 could be a hydroxy derivative of M2. Further reaction of M1 could involve oxidation of the 5'-hydroxymethyl group, giving a 5'carboxy compound related to DMeCP with a molecular weight of 274. In fact, we have recently isolated such a compound from mouse urine [18]. Mass fragmentation did not allow us to determine the location of the proposed hydroxy group on M₁ or on the material of molecular weight 274, but n.m.r. studies, which have not yet been performed, should give the structures of the compounds. Since M₁ and TMeP have similar ultraviolet absorption spectra, the hydroxylation may be on the 4 or 8 methyl group rather than directly on one of the rings. Hydroxylation of a ring position would be expected to show large spectral shifts when compared with TMeP.

DMeCP was not further metabolized in vitro by mouse liver homogenate or the 10,000 g supernatant fraction, or by kidney tissue fractions. This lack of reaction is expected for a metabolic end product. It is not yet clear whether M₂ is excreted in urine after oral administration of TMeP, or whether M₂ is completely transformed to DMeCP under in vivo conditions. Results obtained after feeding mice [³H]TMeP indicate that M₁ is not excreted in mouse urine (B. Mandula and M. Pathak, unpublished observations).

In the reaction of TMeP with liver fractions in vitro, only ether-extractable substances were studied further, although up to 10 per cent of total radio-activity remained in the aqueous phase. This non-volatile, nonextractable material could be either: (a) TMeP metabolites; or (b) reduced pyridine nucleotides, possibly formed in a dehydrogenase reaction.

In our studies of TMeP metabolism, we have not found significant differences among species. Both mice and humans receiving oral TMeP excrete DMeCP. In addition, in vitro studies with guinea pig and mouse liver strongly suggest that livers from both animals form M₁ and M₂ from TMeP.

Urine specimens from mice and humans given 8-MOP orally show at least two ether-extractable fluorescent metabolites and little undegraded 8-MOP (B. Mandula and M. Pathak, unpublished observations). Therefore, the compound is clearly metabolized in vivo. However, when 8-MOP was incubated in vitro with liver fractions under mixed-function oxidase conditions, we were unable to detect substantial formation of products. Several possible explanations may account for these negative results with 8-MOP: (a) the liver may not be the main site of metabolism of 8-MOP; (b) the incubation conditions may not fulfill the requirements for the reactions including the activation or solubilization of the substrate; (c) product or substrate inhibition may prevent the reaction of 8-MOP proceeding to any great extent; (d) products may be unstable or nonether-extractable; or (e) products may lack both fluorescence and a typical furocoumarin absorption spectrum. The first two possibilities seem less likely than the remaining three possibilities, which could be investigated with radioactive 8-MOP.

Failure to detect products from 8-MOP or TMeP after incubation with guinea pig epidermis could indicate that the appropriate enzyme system is either absent or present in amounts too low to be detected by our assay. Although benzpyrene hydroxylase activity (cytochrome P-448) at low levels has been observed in skin from humans [21], rats [22,23] and mice [24,25], we are unaware of any evidence demonstrating cyctochrome P-450 mixed-function oxidases in skin.

The observation that TMeP is rapidly transformed in vitro by the liver but not by the epidermis may be related to therapeutic differences seen in topical and systemic photochemotherapy. Recent unpublished studies in our laboratory have indicated that in man topically applied TMeP at low concentrations (1-5 $\mu g/2.5 \text{ cm}^2$) is very photoreactive and induces marked skin photosensitization reactions (e.g. erythema, edema and cell death). On the other hand, orally administered TMeP (40-120 mg/70 kg) is significantly less active than topical TMeP, orally administered 8-MOP (40 mg/70 kg) or topical 8-MOP $(1-5 \mu g/2.5 \text{ cm}^2)$. The decreased TMeP reactivity in vivo could be related to its rapid biotransformation and excretion. DMeCP, one of the major metabolites of TMeP, is apparently inactive and nonphotosensitizing when tested on guinea pig skin [18]. Recent unpublished clinical experiments showed that topical TMeP cleared psoriasis, whereas orally administered TMeP was ineffective in the same patients. Additional studies are needed, however, to understand the differences in the photoreactivity of 8-MOP and TMeP after oral administration.

Acknowledgements—We acknowledge Dr. Gerald Dudek and Ms. Linda Sher for their invaluable technical help.

This investigation was partially supported by a research grant from the GTE Sylvania Lighting Products Group.

REFERENCES

- J. A. Parrish, T. B. Fitzpatrick, L. Tanenbaum and M. A. Pathak, New Engl. J. Med. 291, 1207 (1975).
- K. Wolff, T. B. Fitzpatrick, J. A. Parrish, F. Gschnait,
 B. Gilchrest, H. Honigsmann, M. A. Pathak and L. Tanenbaum, Archs Derm. 112, 943 (1976).
- 3. G. Weber, Br. J. Derm. 90, 317 (1974).
- B. Gilchrest, J. A. Parrish, L. Tanenbaum, H. A. Haynes and T. B. Fitzpatrick, *Cancer*, *Philad.* 38, 683 (1976).
- T. B. Fitzpatrick, J. A. Parrish and M. A. Pathak, in Sunlight and Man: Normal and Abnormal Photobiological Responses (Eds M. A. Pathak, L. C. Harber, M. Seiji and A. Kukita), pp. 783-93. University of Tokyo Press, Tokyo (1974).
- W. L. Morison, J. A. Parrish and T. B. Fitzpatrick, J. Invest. Derm. 67, 561 (1976).
- F. Dall'Acqua, S. Marciani, D. Vedaldi and G. Rodighiero, Fedn Eur. Biochem. Soc. Lett. 27, 192 (1972).
- L. Musajo, F. Bordin and R. Bevilacqua, *Photochem. Photobiol.* 6, 927 (1967).
- 9. R. S. Cole, Biochim. biophys. Acta 217, 30 (1970).
- F. Dall'Acqua, S. Marciani, L. Ciavetta and G. Rodighiero, Z. Naturf. (B) 26, 561 (1971).
- M. A. Pathak and D. M. Krämer, *Biochim. biophys. Acta* 195, 197 (1969).
- H. P. Baden, J. M. Parrington, J. D. A. Delhanty and M. A. Pathak, *Biochim. biophys. Acta* 262, 247 (1972).
- E. Ben-Hur and M. M. Elkind, *Biochim. biophys. Acta* 331, 181 (1973).
- C. V. Hanson, C. K. J. Shen and J. Hearst, Science, N. Y. 193, 62 (1976).
- 15. D. Averbeck, P. Chandra and R. K. Biswas, *Radiat. environ. Biophys.* 12, 241 (1975).
- J. F. Walter, J. J. Voorhees, W. H. Kelsey and A. Duell, *Archs Derm.* 107, 861 (1973).
- M. A. Pathak, D. M. Krämer and T. B. Fitzpatrick, in Sunlight and Man: Normal and Abnormal Photobiological Responses (Eds M. A. Pathak, L. C. Harber, M. Seiji and A. Kukita), pp. 335-68. University of Tokyo Press, Tokyo (1974).
- B. B. Mandula, M. A. Pathak and G. Dudek, Science, N.Y. 193, 1131 (1976).
- O. H. Lowry, N. J. Rosebrough, A. L. Farr and R. J. Randall, J. biol. Chem. 193, 265 (1951).
- C. J. Mannering, in Fundamentals of Drug Metabolism and Drug Disposition (Eds B. N. La Du, H. G. Mandel and L. L. Way), pp. 206-52. Williams & Wilkins, Baltimore (1971).
- W. Levin, A. H. Conney, A. P. Alvares, I. Merkatz and A. Kappas, Science, N. Y. 176, 419 (1972).
- D. R. Bickers, J. Eiseman, A. Kappas and A. P. Alvares, Biochem. Pharmac. 24, 779 (1975).
- D. R. Bickers, A. Kappas and A. Alvares, J. Pharmac. exp. Ther. 188, 300 (1974).
- F. J. Wiebel, J. C. Leutz and H. V. Gelboin, Archs Biochem. Biophys. 154, 292 (1973).
- F. J. Wiebel, J. C. Leutz and H. V. Gelboin, J. Invest. Derm. 64, 184 (1975).